

ANGULAR HYDROXYLATION OF POLYCYCLIC KETONES
USING BENZENESELENINIC ANHYDRIDE¹

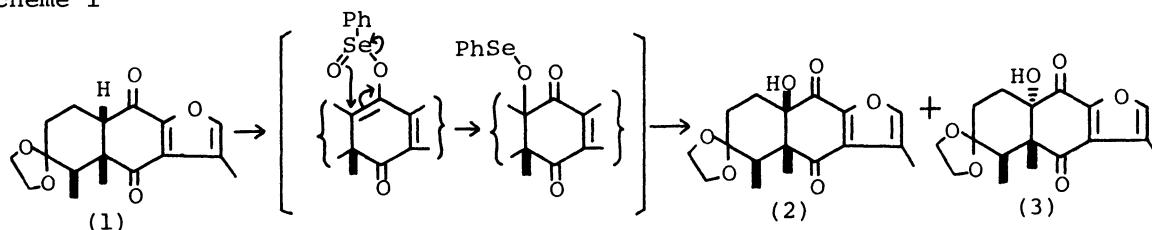
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Introduction of a hydroxyl group into the angular position of tricyclic ketones (1, 4a, 4b, and 6) was accomplished by the use of benzeneseleninic anhydride, the products being obtained in a good yield.

Barton *et al.* reported *ortho*-hydroxylation of phenols using benzeneseleninic anhydride,² although a number of alicyclic alcohols were oxidised by this reagent to carbonyl derivatives in a high yield,³ and the corresponding enones were obtained in some cases by further oxidation.⁴ We report here that introduction of a hydroxyl group into the angular position of tricyclic ketones (1, 4a, 4b, and 6) by benzeneseleninic anhydride was accomplished, and the products were obtained in a good yield, without the formation of corresponding enones.

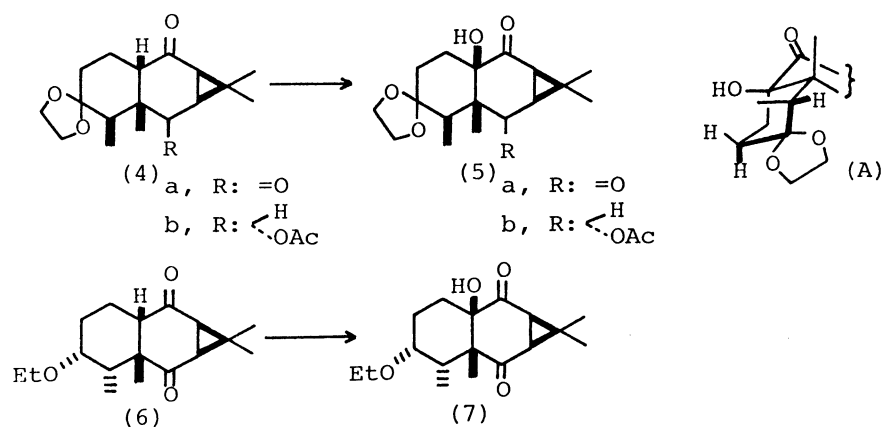
Scheme 1



A solution of 3,3-ethylenedioxy-6,9-dioxofuranoeremophilane⁵ (1) and benzeneseleninic anhydride in toluene was heated under reflux for 4 hr to give 1, 2, and 3, in 24, 57, and 17% yield, respectively. Structure of the major product (2), mp 130-132°, was confirmed from its spectral data [MS: M^+ 320; IR: 3520 cm^{-1} ; UV: $\lambda_{\text{max}}^{\text{EtOH}}$ 243 and 304 nm; NMR δ : 1.17 (3H, d, $J=7$ Hz, 4- CH_3), 1.19 (3H, s, 5- CH_3), 2.30 (3H, 11- CH_3), 2.81 (1H, dq, $J=7, 1$ Hz, 4-H)]. In ^{13}C -NMR spectra of 1 and 2, a doublet signal of 10-C at δ 53.7 in 1 changed to a singlet signal at δ 80.6 in 2 due to introduction of a hydroxyl group. In ^1H -NMR spectrum of 2, 4-H appeared as a quartet signal at δ 2.81, with long-range coupling ($J=1$ Hz) with 2-H attributed to the "W" arrangement. From these data, the structure of 2 was found to be 3,3-ethylenedioxy-10 β -hydroxy-6,9-dioxofuranoeremophilane, shown by the non-steroidal conformation (A). Structure of the minor product (3), mp 195-199°, was confirmed from its spectral data [MS: M^+ 320; IR: 3490 cm^{-1} ; UV: $\lambda_{\text{max}}^{\text{EtOH}}$ 243 and 302 nm; NMR δ : 1.16 (3H, d, $J=7$ Hz, 4- CH_3), 1.27 (3H, s, 5- CH_3), 2.25 (3H, 11- CH_3), 2.85 (1H, q, $J=7$ Hz, 4-H)].

A solution of tricyclic ketones (4a, 4b, and 6), which were prepared from the major adduct of 3-ethoxy-1,3-pentadiene and 3-carene-2,5-dione,⁶ and benzeneseleninic anhydride in toluene, (for 4a in chlorobenzene), was heated under reflux for 4-5 hr. 10 β -Hydroxy compounds, 5a, mp 148-150°, 5b, mp 170.5-172°, and 7, mp 155-157°, were obtained in 73, 80, and 70% yields (from converted 4a, 4b, and 6), respectively. Structures of 5a, 5b, and 7 were confirmed from their spectral data. Hydroxyl absorption band in IR spectra of 5a, 5b, and 7 appeared at 3450, 3450, and 3380 cm⁻¹, respectively. In ¹³C-NMR spectra of 5a, 5b, and 7, the chemical shift of 10-C appeared as a singlet signal at δ 79.7, 78.1, and 79.9, respectively. In ¹H-NMR spectra of 5a and 5b, 4-H appeared as a quartet signal at δ 2.58 and 2.61, respectively, with long-range coupling with 2-H, attributed to the "W" arrangement. From these spectral data, the stereoformulae of 5a and 5b should be shown by the non-steroidal conformation (A) (Scheme 2).

Scheme 2



Benzeneseleninic anhydride is an effective reagent for angular hydroxylation of a variety of polycyclic ketones which are enolizable in the angular position.

References and Note

1. Studies on Terpenoids and Related Alicyclic Compounds XIX. Part XVIII. K. Yamakawa and T. Satoh, Chem. Pharm. Bull. (Tokyo), in press.
2. D.H.R. Barton, S.V. Ley, P.D. Magnus, and M.N. Rosenfeld, J. Chem. Soc. Perkin I, 567 (1977).
3. D.H.R. Barton, A.G. Brewster, R.A.H.F. Hui, D.J. Lester, and S.V. Ley, J.C.S. Chem. Comm., 952 (1978).
4. D.H.R. Barton, D.J. Lester, and S.V. Ley, J.C.S. Chem. Comm., 130 (1978).
5. K. Yamakawa and T. Satoh, Chem. Pharm. Bull. (Tokyo), 26, 3704 (1978).
6. K. Yamakawa, R. Sakaguchi, and T. Satoh, Symposium papers, 20th Symp. Chem. Natural Products, Sendai, 139-146 (1976).

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